154. The Reaction Mechanism of Chromanol-Ring Formation Catalyzed by Tocopherol Cyclase from *Anabaena variabilis* KÜTZING (Cyanobacteria)

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Incubation of the synthetic ¹⁸O-labelled phytyl-hydroquinone (O^{4} -¹⁸O)-2 with the tocopherol cyclase from Anabaena variabilis Kützing (Cyanobacteria) in D₂O furnished the doubly labelled γ -tocopherol, (2R,3S,4'R,8'R)- $(1^{-18}O,3^{-2}H)$ -1. The chirality at C(3) was determined by two independent routes providing interlocking evidence that the enzyme-catalyzed ring closure proceeds by si-protonation of the double bond of 2 and concomitant re-attack of the phenolic O-atom.

Introduction. – Recently, we identified a new enzyme in the blue-green algae Anabaena variabilis KÜTZING (Cyanobacteria) which catalyzes the formation of γ -tocopherol (1) from the phytyl-hydroquinone 2 (Scheme 1). It was shown that the cyclization was stereospecific and could be driven to nearly quantitative substrate turnover under strictly defined conditions such as incubating the 2,6-di-O-methyl- β -cyclodextrin complex of 2 in the presence of ascorbic acid with spheroplasts prepared by lysozyme treatment of intact Anabaena cells [1].

The mechanism of this cyclization, in principle, could proceed through different intermediates, as outlined in *Scheme 2. Route A* involves the stereospecific hydration of the (E)-double bond of **2** followed by cyclization of $3 (\rightarrow 1)$ under retention of configuration at the tertiary alcohol. The latter reaction resembles the well established acid-cata-

lyzed ring-closure of 3 in organic solvents, vide infra [2]. On the other hand, the reverse addition of H_2O (or any other species like HS-enzyme) to the double bond of 2 would yield an alcohol 4 (or thio adduct) which could cyclize (\rightarrow 1) under inversion of configuration by attack of the phenolic O-atom (Route B). The third possibility, Route C, simply involves the stereospecific addition of the phenol to the protonated double bond of 2. Route A is distinguished from Routes B and C by the elimination of the phenolic O-atom and Routes B and C are distinct with respect to the configuration at C(3) of 1, if the incubation of 2 is done in deuterated buffer, as shown for convenience in Scheme 2.

It is obvious that a decisive experiment can be accomplished if I) an enzyme fraction can be prepared devoid of H_2O , 2) a high deuterium incorporation at C(3) is obtained from deuterated buffer, and methods can be developed to determine the absolute configuration at C(3) of 1, and 3) a precursor specifically ¹⁸O-labelled at C(4) of 2 can be used without exchange of ¹⁸O under conditions of incubation.

Results and Discussion. – To investigate the incorporation of the phenolic O-atom at C(4) of 2, the benzohydroquinone diacetate ($O^{4_{-}18}O$)-5 was prepared according to *Scheme* 3. The H₂O-free, crystalline diazonium tetrafluoroborate 6 of 2,3-dimethylaniline [3] was treated with H₂¹⁸O in THF to yield the isotopically substituted phenol (^{18}O)-7. The

Scheme 3

Ac

$$Ac$$
 $C) R'-Br (9)$
 Ac
 $C) R'BO)-10$
 $C) R'BO$

a) H₂¹⁸O/THF, r.t. →65°; 16%. b) CICH₂OMe, (i-Pr)₂EtN, 40°; 38%. c) l. BuLi, Me₂N(CH₂)₂NMe₂, Et₂O, 2 h; 2. CuBr, 9, Et₂O, −20°, 4 h. d) THF/AcOH/i-PrOH/37% HCl soln., r.t., 4 h; 33% from 8. e) O₂/salcomine, EtOH, r.t. f) Na₂S₂O₄, H₂O/Et₂O, r.t., 4 h. g) Ac₂O/pyridine, Na₂S₂O₄; 61% from 10.

corresponding methoxymethyl ether [4] $(O^{1_-18}O)$ -8 serves as a guide for regiospecific *ortho*-lithiation and subsequent alkylation with phytyl bromide 9 [5] to give (^{18}O)-10 after acid-catalyzed deprotection. Salcomine (= bis(salicylidene)ethylenediimino-cobalt(II)) oxidation [6] [1] and reductive acetylation furnished the desired diacetate ($O^{4_-18}O$)-5 (95% ^{18}O) in 9% overall yield from 7. Enzymatic reactions were done with the corresponding hydroquinone ($O^{4_-18}O$)-2, prepared by LiAlH₄ reduction of ($O^{4_-18}O$)-5, and subsequent formation of its 2,6-di-O-methyl- β -cyclodextrin complex [1] in the presence of ascorbic acid.

To prepare a H_2O -free enzyme fraction, the spheroplasts were cracked by osmolysis at 4° and pH 7.0 and the H_2O -soluble proteins removed by centrifugation, and the resulting pellet was resuspended in buffer. The addition of cold acetone furnished a precipitate which could be washed with cold Et_2O to remove the endogenous tocopherols, ground in a mortar, and dried under high vacuum to give a grey-blue 'acetone powder'. This material was suspended in deuterated buffer in the presence of the non-ionic detergent dodecyl p-maltoside (11) and displayed enzymatic activity (90% conversion $2 \rightarrow 1$) comparable to intact spheroplasts [1]. From an upscaled incubation of 2, mg quantities of the deuterated tocopherol were isolated, and its corresponding methyl ether (3- 2 H)-12 was analyzed by mass and 1 H-NMR spectroscopy. Both methods demonstrated a nearly quantitative deuterium incorporation. Due to specific fragmentation in the MS,

it was evident that only C(3) was monodeuterated, and the ¹H-NMR revealed that just one of the diastereotopic positions at C(3) was deuterated. From incubation of $(O^{4_-18}O)$ -2 under the same conditions in D_2O (see *Scheme 4*), a doubly labelled tocopherol was isolated and its methyl ether $(1^{-18}O, 3^{-2}H)$ -12 prepared for spectroscopical analyses. The ¹H-NMR fully confirmed the former experiment; moreover the M^+ showed ca.83% enrichment of ¹⁸O and ²H (*Fig. 1*). In contrast, no ¹⁸O incorporation into 1 and 12 was observed on incubation of the cyclodextrin complex of unlabelled 2 with the 'acetone powder' solubilized with 11 in a buffer containing $H_2^{-18}O/H_2O$ 1:1, indicating that no exchange of ¹⁸O with the medium had occurred. These results exclude *Route A* (*Scheme 2*) as a mechanism of the enzymatic reaction but leave the choice open between *Routes B* and C.

In view of the fact that on irradiation of the benzylic protons of 12 in a ¹H-NMR experiment the complicated m for the protons at C(3) collaps to a well-defined AB system, which in the sample of enzymatic origin is even more simplified ($\rightarrow s$ at 1.77 ppm), it was decided to prepare deuterated reference probes of known absolute configuration. Two of the many possible strategies to achieve this goal are reported below.

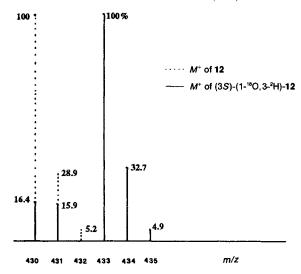


Fig. 1. Mass spectra of $(1^{-18}O,3^{-2}H)$ -12 obtained from tocopherol cyclase treatment of $(O^{4-18}O)$ -2 in D_2O and of the corresponding unlabelled 12

The concept of the 'Basel route' was to make use of γ -tocopheryl acetate (13) as a starting material taking advantage of its easy DDQ (= 2,3-dichloro-5,6-dicyano-1,4benzoquinone = 4,5-dichloro-3,6-dioxocyclohexa-1,4-diene-1,2-dicarbonitrile) oxidation [7] to the didehydro derivative 14 (67% overall yield) from which, on treatment with N-bromoacetamide [8], regioselectively a pair of trans-bromohydrins 15 and 16 was prepared and separated by chromatography. Reduction of 15 with LiEt, BH gave, via the intermediate epoxide 17 (in situ), alcohol 18. When the corresponding triflate 19 (Tf = CF₃SO₂) was treated with LiEt₃BD, the deuterated γ -tocopheryl methyl ether (3-2H)-12 (8% yield by GLC) was isolated by HPLC, besides 14 (65%), 18 (19%), and ca. 2% of benzofurans produced by rearrangement. The same products, but in a different ratio, were obtained from tosylate 20 in a much slower reaction under forced conditions. Keystones of this sequence are the X-ray analyses of bromohydrin 16 and of tosylate 20 derived from alcohol 18. It is interesting to note that the cell unit of 16 contains two individual molecules not related by crystallographic symmetry (Fig. 2). In the crystal of 20, most of the alkyl chain is disordered; perspective drawings of the rotamers A and B obtained from the analysis at 115 K are shown in Fig. 3. These results prove the absolute configuration of all chiral centers in question, in particular the (R)-configuration at C(3)in both cases.

Since the nucleophilic displacement of the homobenzylic triflate group is of significance to the chirality at C(3) of the final product $(3^{-2}H)$ -12, the dideuterated tocopherol derivative (3R,4S)- $(3,4^{-2}H_2)$ -12 was prepared in three steps from 16 (Scheme 5). This compound was also obtained in a 3:5 mixture with (3S,4R)- $(3,4^{-2}H_2)$ -12 (not shown) from 14 by cis-addition of D₂ in the presence of Wilkinson catalyst [9]. It was obvious from the ${}^{1}H$ -NMR spectra that the protons at C(3) and C(4) display a cis-coupling constant of 6.2 Hz in both cases. Almost the same vicinal-coupling constant (6.15 Hz) is observed between H_g at C(3) and H_g at C(4) in the ${}^{1}H$ -NMR of (4R)- $(4^{-2}H)$ -12, prepared in 3 steps

a) Ç6 ^{C5}/ **C8** b) οĭ C29A C30A /C12 C25B C31A С27В C32A C35B C26E **TCS9B** C28B C36B *[*C34B C30B) Fig. 3. X-Ray crystal structure of 20: perspective C32B drawing a) of rotamer A (50% occupancy) and b) of rotamer B (50% occupancy). Arbitrary numbering.

molecule and of b) the other independent molecule. Arbitrary numbering.

Scheme 5

R = (4R,8R)-4,8,12-trimethyltridecyl

a) 1. DDQ, dioxane, reflux; 82%; 2. K_2CO_3 , MeOH, r.t.; 96%; 3. NaH, (MeO)₂SO₂, THF, r.t.; 85%. b) 1. N-Bromoacetamide, THF/H₂O, r.t.; 81%; 2. chromatographic separation, SiO₂. c) 2.5 Equiv. LiEt₃BH, THF, $-78^{\circ} \rightarrow r.t.$; 95–100%. d) TsCl, Et₃N, pyridine, reflux; 65%. e) Tf₂O, 2,6-dimethylpyridine, Et₂O, $-78^{\circ} \rightarrow r.t.$; ca. 100%. f) 20 Equiv. LiEt₃BD, THF, reflux, 5 d. g) 10 Equiv. LiEt₃BD, THF, $-78^{\circ} \rightarrow r.t.$ h) 2.5 Equiv. LiEt₃BD, THF, $-78^{\circ} \rightarrow r.t.$ i) 10 Equiv. LiEt₃BH, THF, $-78^{\circ} \rightarrow r.t.$

Scheme 6

a) 3-Chloroperbenzoic acid, CH₂Cl₂/sat. Na₂CO₃ soln., 0°→r.t.; 82 %. b) LiAlD₄, Et₂O, 0°, 2 h. c) Air; 83 %. d) Na₂S₂O₄, CHCl₃/EtOH/H₂O; TsOH (cat.)/benzene, 80°, 2 h; 87 %. e) (−)-Camphanoyl chloride, pyridine, r.t., 16 h; 96 %; separation on HPLC, see *Fig. 4*. f) LiAlH₄, Et₂O, r.t., 2 h; 94 %. g) NaH, Mel, THF, r.t., 10 min; 96 %.

from 15; however, for this monodeuterated compound, the *trans*-coupling constant of 6.85 is also detected. This clearly indicates that removal of the triflate group at C(3) proceeded with inversion of configuration. Together with the X-ray structures of 16 and 20 taken as structural references for the sequence, the (S)-configuration at C(3) of $(3-^2H)$ -12 is established, and the compound is suitable for comparison with the sample produced by the enzyme 'tocopherol cyclase'.

The 'Zürich route' takes into account common knowledge [2] concerning the stereospecific ring closure of the tertiary alcohol 3 yielding 1 under retention of configuration (see above, Scheme 2, Route A), and the fact that (R,R,R)- and (S,R,R)-tocopheryl camphanates can be separated by chromatography. Thus, the procedure is very simple $(Scheme\ 6)$.

Epoxidation of the diacetate 5 gave a mixture of diastereoisomeric epoxides 21 and 22 that was immediately treated with LiAlD₄ and oxidized to yield the two tertiary alcohols (2'S,3'R)-(2'-2H)- and (2'R,3'S)-(2'-2H)-23. Without separation, the latter were reduced

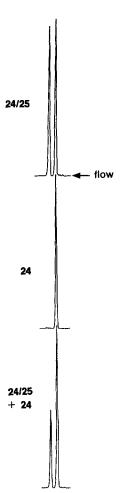


Fig. 4. HPLC analysis of the diastereoisomeric (-)-camphanates 24/25 (obtained from (2RS,4'R,8'R)-y-tocopherol), of 24 (obtained from (2R,4'R,8'R)-y-tocopherol), and of a mixture of both. Conditions: Spherisorb-3S-CN column, hexane/dioxane 96:4, 250 × 4.0 mm, 62 bar, r.t., flow 1.0 ml/min, 220 nm.

and cyclized [2] to furnish the two γ -tocopherols with opposite configurations at C(2) and C(3). The corresponding camphanates (2R,3S)- $(3^{-2}H)$ -24 and (2S,3R)- $(3^{-2}H)$ -25 were baseline-separated on HPLC and identified by HPLC comparison with an unlabelled mixture 24/25, prepared from synthetic 2-ambo- γ -tocopherol ((2RS)-1), and coinjection of 24/25 with unlabelled 24, prepared from natural (2R)-1 (Fig. 4). Accordingly, the faster-eluting compound proved to be the (2R)-configurated isomer 24. Thus, the corresponding (2R,3S)- $(3^{-2}H)$ -24 was hydrolyzed and methylated to finally give the desired reference sample (3S)- $(3^{-2}H)$ -12 (Scheme 6).

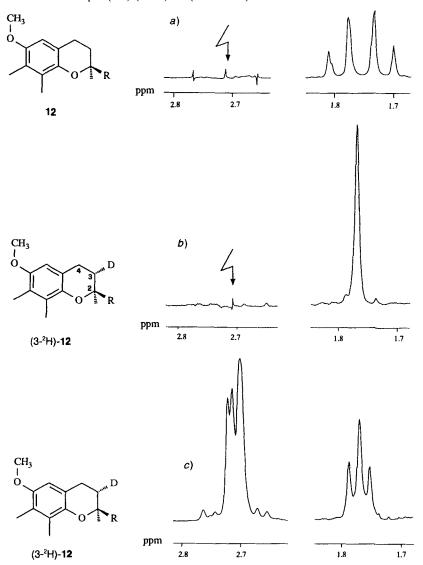


Fig. 5. Partial 400-MHz 1 H-NMR spectra (region of H–C(3) and H–C(4)) of a) unlabelled 12, selectively decoupled, and of b) c) deuterated (3S)-(3- 2 H)-12, with and without decoupling, respectively

Spectroscopic comparison of the two synthetic probes of (3-2H)-12 with the compound of enzymatic origin for a partial 'H-NMR-spectrum (see Fig. 5), revealed the identity of the three samples in every respect, and thus firmly establishes the mechanism of the enzymatic reaction catalyzed by 'tocopherol cyclase' to operate by si-protonation of the double bond of 2 and concomitant re-attack of the phenolic O-atom (Route C in Scheme 2). Whether the transition state of the cyclization is early or late on the reaction coordinate will be investigated by testing the binding profile of different possible transition-state analogues to the purified enzyme [10].

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Experimental Part

General. See [1]. THF was freshly distilled from Na/benzophenone under Ar. H₂¹⁸O (≥ 99.5% ¹⁸O) was purchased from Ventron, natural trans-phytol from Eisai Co., Ltd., Tokyo, Japan; lot No. 590'222, assay: 87.7% by GLC. The latter was further purified by column chromatography on silica gel (1 kg for 25 g of phytol, hexane/Et₂O 1:1). The residue was dissolved in hexane and washed twice with NaHCO₃ soln., the hexane soln. evaporated, and the residue distilled (bulb-to-bulb, ca. 120°/0.1 mm) to give trans-phytol (= (E,7R,11R)-3.7.11.15-tetramethylhexadec-2-enol), purity $\geq 98\%$ (GLC). This was subsequently condensed with 2.3-dimethylhydroquinone in HCOOH [11] [12] to yield 2-ambo-y-tocopherol ((2RS,4'R,8'R)-1; see optimized procedure below). Natural y-tocopherol was isolated from the commercially available tocopherol concentrate 'd-mixed tocopherols' (Bizen Chemical Co., Ltd., Kumayama, Akaiwa, Okayama, 709-07, Japan; composition according to GLC: γ -tocopherol (40%), δ -tocopherol (23%), α -tocopherol (8%), β -tocopherol (0.8%)) by column chromatography (3.5 kg Merck silica gel 60, particle size 0.063-0.200 mm for 60 g 'd-mixed tocopherols', hexane/AcOEt 88:12) to yield ca. 97% pure material which was acetylated (Ac₂O/pyridine), chromatographed (SiO₂, hexane/ AcOEt 95:5), and distilled (bulb-to-bulb): purity of the acetate 13 ca. 98 % (GLC). The corresponding methyl ether derivative 12 was shown to be stereoisomerically pure (>99.5% by HPLC) [13]. In addition: TLC: Detection of substances by UV (254/366 nm) and ammonium molybdate/Ce(SO₄)₂ in H₂O/H₂SO₄ with subsequent heating. GLC: PS-086 capillary column, 14 and 15 m. Supercritical fluid chromatography (SFC): Instrument Lee Scientific, model 600 (Dionex, Salt Lake City, UT, USA), FID at 380°, fused-silica capillary colum coated with biphenyl-30, length 10 m, internal diameter 50 μm, mobile phase CO₂, 100°, 0.2-0.75 g ml⁻¹, linear ramp at 0.01-0.005 g·ml⁻¹·min⁻¹ [16]. Melting points (uncorrected): Büchi-510 apparatus. Optical rotations: Perkin-Elmer-241 polarimeter, c in g/100 ml.

'Acetone Powder' Containing Tocopherol Cyclase. Spheroplasts were obtained from intact cells of Anabaena variabilis KÜTZING (Cyanobacteria) according to [1]. The spheroplasts (52 g) were suspended twice in 300 ml of ice-cold phosphate buffer (10 mm, pH 7.0) and centrifuged at 1500 g/15 min. The resulting pellet was suspended in 40 ml of the same phosphate buffer at 4° , 11 of acetone (-20°) was added to the vigorously stirred medium, and the insoluble proteins were collected by centrifugation at 1500 g/2 min. The org. phase was removed and the protein pellet extracted for the lipids with 500 ml of ice-cold Et_2O (10 s vigorous shaking by hand). Finally the suspension was filtered through cotton and the residue dried under high vacuum for 3 h. After grinding in a mortar, 2.1 g of a slightly blue 'acetone powder' were obtained, which could be stored at -80° for ca. 2 months without loosing more than 20% of the enzymatic activity. According to HPLC analysis, extracts of the 'acetone powder' contained no traces of tocopherols.

Tocopherol-Cyclase Activity of the 'Acetone Powder'. The 'acetone powder' (4.5 g) was suspended at 4° in buffer (pH 7.0, 200 ml) containing dodecyl p-maltoside (11; 15 mm), potassium phosphate (100 mm), and pL-dithiothreitol (2 mm). After stirring for 1 h at 4° with a magnetic bar, the suspension was centrifuged at 100.000 g for 1 h

and the supernatant (186 ml) used for incubations as follows. Buffer (40 μ l) containing the substrate inclusion complex [I] was added to the solubilisate (1 ml) such that the following final concentrations were obtained: 96 μ m for 2 (40 μ g), 1.8 mm for 2,6-di-O-methyl- β -cyclodextrin, and 10 mm for ascorbic acid. Incubation was performed at 35° for 15 h to yield, after usual workup [I] and detection by HPLC (internal standard/fluorescence det.), 36 μ g (90%) of 1.

Incubation of **2** *with Solubilized 'Acetone Powder' in D*₂O. Both substrate buffer and 'acetone powder' (4 g) were prepared and solubilized, respectively, in D₂O (≥ 99.8 % D). Incubation of **2** (7.5 mg, 18.0 μmol) furnished (3*S*)-(3-²H)-**1** (6.29 mg, 83.9 %) which was immediately methylated by treatment with an excess of NaH/MeI in abs. THF [1]. Workup and purification on TLC (silica gel, hexane/CH₂Cl₂ 2:1) gave γ-tocopheryl methyl ether (3*S*)-(3-²H)-**12** (5.6 mg, 71 %), pure both by GLC (97.5 %) and HPLC (99.4 %). ¹H-NMR (400 MHz): 6.43 (*s*, H-C(5)); 3.76 (*s*, MeO); 2.72 (br. *d*, *J* = 7.0, 2 H-C(4)); 2.13 (*s*, Me-C(7)); 2.12 (*s*, Me-C(8)); 1.78 (*t*, *J* = 7.0, H-C(3)); 1.61–1.02 (*m*, 21 H); 1.25 (*s*, Mc-C(2)); 0.88–0.84 (*m*, 12 H). MS: 433 (6), 432 (36), 431 (100, *M* ⁴⁺), 206 (12, [*M* − C₁₆H₃₃]⁺), 166 (18), 165 (100, [*M* − C₁₉H₃₇]⁺), 164 (27).

Incubation of $(O^4-^{18}O)-2$ with Solubilized 'Actone Powder' in D_2O . 'Acetone powder' (1.7 g) and substrate buffer were prepared in D_2O ($\geq 94\%$ D) and $(O^4-^{18}O)-2$ (2.0 mg, 4.8 µmol) incubated at 35° for 15 h. After workup and purification, (3S)-(1-¹⁸O,3-²H)-1 (0.4 mg, 20%) was isolated and directly methylated as described [1] to yield (3S)-(1-¹⁸O,3-²H)-12, pure by HPLC (97.7%). MS: 435 (5), 434 (33), 433 (100, M^{+}), 431 (16), 430 (16), 208 (5, $[M-C_{16}H_{33}]^+$), 168 (8), 167 (56, $[M-C_{19}H_{37}]^+$), 166 (14), 165 (14), 164 (3).

Incubation of 2 with 'Acetone Powder' Solubilized in $H_2^{18}O$. A control incubation of 2 (0.5 mg, 2 µmol) in substrate buffer/solubilized 'acetone powder', both prepared in $H_2^{18}O$ (50% ^{18}O), gave γ -tocopherol (1) and its corresponding methyl ether 12, purified by HPLC. The MS of this sample was completely identical both with respect to peak distribution and peak integration with the MS of unlabelled 12. MS: 432 (5), 431 (29), 430 (100, M^+), 205 (9, $\{M - C_{16}H_{33}\}^+$), 166 (9), 165 (75, $\{M - C_{19}H_{37}\}^+$), 164 (18).

2,3-Dimethylbenzene-1-diazonium Tetrafluoroborate (6°). Slowly 2,3-dimethylaniline (12.1 g, 0.1 mol) was added to a mixture of 37% aq. HCl soln. (25 ml, 0.3 mol) and H₂O (50 ml) at 10°. After 10 min, a soln. of NaNO₂ (7 g, 0.1 mol) in H₂O (100 ml) was added dropwise to the mechanically stirred precipitate at -5° . After 1 h stirring at 0°, NaBF₄ (11.6 g, 0.102 mol) in H₂O (100 ml) was added in 30 min. Then the beige precipitate was filtered, washed with cold (0°) H₂O (2 × 25 ml), cold (-20°) EtOH (4 × 25 ml), and Et₂O (8 × 50 ml), and dried at 0°/1 Torr: 6 (12.0 g, 55%). Beige crystals. M.p. *ca.* 80° (dec.). Anal.: H₂O-content < 0.1%.

2,3-Dimethyl(^{18}O)phenol ((^{18}O)-7). A suspension of **6** (4.0 g, ca. 18 mmol) in THF (15 ml)/H₂¹⁸O (1 ml; ≥ 99.5% ^{18}O) was stirred at r.t. for 1 h, then at 35° for 2 h, and at 65° for 30 min. Then the mixture was poured into hexane/Et₂O 3:1 (200 ml) and extracted with 2M NaOH (4 × 100 ml). Neutralization of the aq. phase with 37% aq. HCl soln., extraction with Et₂O (3 × 100 ml), evaporation, and chromatography (SiO₂ (60 g), hexanc/AcOEt 4:1) gave (^{18}O)-7 (350 mg, 16%). Beige crystals. M.p. 69.5°. GLC Purity 99.6%. 1 H-NMR (250 MHz): 2.16, 2.27 (2s, Me–C(2), Me–C(3)); 4.59 (s, OH); 6.63, 6.76 (2d, $J \approx 8$, H–C(4), H–C(6)); 6.97 (t, $J \approx 8$, H–C(5)). MS: 124 (74, M^+ , 96% ^{18}O), 109 (100), 91 (25), 79 (26), 77 (36), 57 (26).

 $(O^{1-18}O)$ -1-(Methoxymethoxy)-2,3-dimethylbenzene $((O^{1-18}O)$ -8). To a cold (0°) soln. of (^{18}O) -7 (330 mg, 2.6 mmol, 96% ^{18}O) and (i-Pr)₂EtN (5 ml, 30 mmol) in CH₂Cl (5 ml) was added 6.1 m ClCH₂OMe (4 ml, 24.4 mmol in AcOMe; *in situ* prepared [4]). After 5 h stirring at 40°, 25% aq. NH₃ soln. (5 ml) was added at r.t. Then the mixture was poured into Et₂O (50 ml) and the org. phase washed with 3n HCl, sat. NaHCO₃ soln., and brine, dried (Na₂SO₄), and evaporated: 300 mg of yellow oil. Bulb-to-bulb distillation at *ca.* 50°/0.1 mbar afforded pure $(O^{1-18}O)$ -8 (170 mg, 38%). Colourless oil. GLC Purity 99.6%. 1 H-NMR (250 MHz): 2.18, 2.27 (2s, Me-C(2), Me-C(3)); 3.49 (s, MeO); 5.18 (s, OCH₂O); 6.82, 6.92 (2d, $J \approx 7$, H-C(4), H-C(6)); 7.05 (t, $J \approx 7$, H-C(5)). MS: 168 (15, M⁺, ca. 100% ^{18}O), 138 (8), 123 (5), 105 (6), 91 (5), 77 (7), 45 (100).

2,3-Dimethyl-6-phytyl (^{18}O) phenol ((^{18}O)-10). BuLi (1.6M in hexane; 0.8 ml, 1.3 mmol) was added to a cold (0°) soln. of ($O^{1-18}O$)-8 (170 mg, 1 mmol) in N, N, N, N-tetramethylethylenediamine (0.2 ml, 1.3 mmol) and Et₂O (2 ml). After 2 h, CuBr (15 mg) and a soln. of phytyl bromide (9; 0.5 g, 1.4 mmol; prepared from *trans*-phytol according to [5]) in Et₂O (2 ml) was added at -20° . After 4 h at -20° , the mixture was mixed with Et₂O (50 ml), washed with 2N H₂SO₄/MeOH 1:1, H₂O/MeOH 1:1, and brine to give, after evaporation, a yellow oil (450 mg) which was treated at r.t. with freshly prepared THF/AcOH/i-PrOH/37% aq. HCl soln. 10:10:10:1 (5 ml). After 4 h, the mixture was diluted with hexane (50 ml) and washed with 1/2-sat. NaHCO₃ soln./MeOH 1:1 (30 ml). Usual evaporation and chromatography (SiO₂, hexane/AcOEt 19:1) yielded (^{18}O)-10 (160 mg, 33%). Yellow oil. GLC Purity: (E) + (Z) 83.5%, (E)/(Z) 98:2. MS: 402 (12, M), 177 (37), 149 (13), 137 (100, 96% ^{18}O).

Preparation of 6 according to a general procedure described in [3]; it decomposes slowly at r.t. and should be stored at -20°, if necessary.

 $(O^{4-18}O)$ -2,3-Dimethyl-5-phytyl-1,4-hydroquinone Diacetate (= $(O^{4-18}O)$ -2,3-Dimethyl-5-phytylbenzene-1,4-diyl Diacetate; $(O^{4-18}O)$ -5). A soln. of (^{18}O) -10 (150 mg, 0.31 mmol; GC: 83.5%) and salcomine [6] (15 mg, 0.05 mmol) in EtOH (5 ml) was stirred under O_2 (1.2–1.4 atm) for 4 h at r.t. Evaporation and chromatography (SiO₂, toluene) gave a yellow oil ($(O^{4-18}O)$ -2,3-dimethyl-5-phytyl-1,4-benzoquinone, 95 mg; HPLC purity: 88%), which was dissolved in Et₂O (5 ml) and treated with a soln. of $Na_2S_2O_4$ (300 mg, 1.7 mmol) in H_2O (3 ml). After stirring for 4 h at r.t., the mixture was extracted with Et_2O (75 ml) and the extract washed with brine and evaporated. The remaining oil was acetylated with Ac_2O (0.5 ml) and pyridine (4 ml) in the presence of $Na_2S_2O_4$ (200 mg). Usual workup gave a yellow oil (120 mg) which was chromatographed twice on SiO_2 (30 g, toluene; 30 g, hexane/AcOEt 6:1): $(O^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: (E) + (Z) 95.4%, (E)/(Z) 96:4. ¹H-NMR (250 MHz): $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 61%). Slightly yellow oil. GLC Purity: $(D^{4-18}O)$ -5 (60 mg, 7 CH₂(1)); 5.2 (t, $D^{4-18}O$ -5 (7), Me-C(1)); 2.31, 2.32 (2s, 2 Ac); 3.16 (d, $D^{4-18}O$ -7 (2h₂(1)); 5.2 (t, $D^{4-18}O$ -8 (100, 95%) 18 (100, 95%) 18 (100, 95%) 18 (100, 95%) 18 (100, 95%) 18 (100, 95%) 18 (100, 95%) 18 (100, 95%) 18 (1

 $(2\,\mathrm{R})$ -6-Methoxy-2,7,8-trimethyl-2-[$(4\,\mathrm{R},8\,\mathrm{R})$ -4,8,12-trimethyltridecyl]-2H-1-benzopyran (= $(2\,\mathrm{R},4'\,\mathrm{R},8'\,\mathrm{R})$ -3,4-Didehydro-6-O-methyl- γ -tocopherol; **14**). The soln. of $(2\,\mathrm{R},4'\,\mathrm{R},8'\,\mathrm{R})$ - γ -tocopheryl acetate (**13**; obtained from natural sources; 45.90 g), and DDQ (34.05 g) in dioxane (600 ml) was stirred under refluxed for 6 h [7]. After stirring for additional 15 h at r.t., the mixture was filtered and the residue washed with hexane (200 ml). The combined filtrate was diluted with hexane (500 ml) and washed twice with H₂O (1000 ml), the H₂O phase extracted with hexane (500 and 1000 ml), the combined org. phase dried (MgSO₄) and evaporated, and the crude black oil (ca. 52 g) purified by chromatography (silica gel (2 kg), hexane/Et₂O 9:1; TLC: R_f 0.39 (**13**), 0.33 (didehydro acetate)): 37.65 g (82%) of $(2\,\mathrm{R},4'\,\mathrm{R},8'\,\mathrm{R})$ -3,4-didehydro- γ -tocopheryl actate [7] as a pale yellow oil, purity (GLC) 98.7–98.8% 2).

To a soln. of this material (22.84 g, 50.0 mmol) in MeOH (220 ml), K_2CO_3 (3.4 g, ca. 25 mmol) was added at r.t. After ca. 1 h of vigorous stirring (TLC control (CH₂Cl₂): R_f 0.65 (didehydro acetate), 0.45 (didehydrophenol)), the brown mixture was evaporated, the residue dissolved in Et₂O (300 ml) and the soln. washed with sat. NaCl soln. and H₂O (100 ml each), dried (MgSO₄), and evaporated: 19.90–20.70 g (96–100%) of (2R,4'R,8'R)-3,4-didehydro-y-tocopherol [7] as a brownish oil, purity (GLC) 97.7% ²), which was used without further purification in the next step.

The soln. of this material (19.98 g, 48.2 mmol) in THF (75 ml) was added dropwise to the stirred suspension of NaH (1.45 g, 60.4 mmol, 1.25 equiv.; prepared from 55 or 80 % NaH/mineral oil by washing with hexane several times) in THF (100 ml) at r.t. during 5 min. After completion of H_2 evolution (1.5–2 h), Me₂SO₄ (6.00 ml, 63.0 mmol, 1.3 mol-equiv.) was added dropwise and stirring continued for 2.5 h (TLC control, (hexane/AcOEt 9:1): R_f (14) 0.59, R_f (didehydrophenol) 0.29). Then 2N NaOH (50 ml) was added carefully, the mixture stirred for 1 h and then extracted with E_{12} O (250, 200, 100 ml), the combined extract washed with H_{20} O (100 ml), dried (MgSO₄), and evaporated and the crude yellowish oil (ca. 20 g) chromatographed (silica gel (220 g), hexane/AcOEt 195:5): 17.46–19.15 g (85–93%) of 14 [7]. Pale yellow oil. Purity (GLC) 97.6. [α] $_{589}^{20}$ = -44.1 (c = 1.0, CHCl₃). The anal. data given below resulted from a distilled (250°/1–2 mbar, bulb-to-bulb distillation, $618 \rightarrow 521$ mg) sample which, however, contained both the (2R)- and the (2S)-stereoisomer due to thermal epimerization at C(2)²). Colourless oil. Purity (GLC) 98.2%. IR (film): 2927s, 2867m, 1462s, 1120m, 1100m, 845ss. H-NMR (250 MHz, CDCl₃): 0.84 (d, d = 7, 2 d d d (d) = 7, 2 d d d (d) = 7, 2 d d d (d) = 7, 2 d d d (d), 3.76 (d), 40–C(2)); 2.12 (d), Me–C(3), 0.84 (d), 4 = 7, 2 d d d d) = 7, 4 d d d0, 4 = 9.7, 4 d0, 4 = 9.7, 4 d0, 6.37 (d0, arom. H). MS: 428 (3.5, d0, 413 (4, d0, d1), 203 (100, d1), 40, 413 (4, d1), 40, 413 (4)

 $(2\,R,3\,S,4\,R)$ -3-Bromo-3,4-dihydro-6-methoxy-2,7,8-trimethyl-2-[(4R,8R)-4,8,12-trimethyltridecyl]-2H-1-benzopyran-4-ol (15) and (2R,3R,4S)-3-Bromo-3,4-dihydro-6-methoxy-2,7,8-trimethyl-2-[(4R,8R)-4,8,12-trimethyltridecyl]-2H-1-benzopyran-4-ol (16). N-Bromoacetamide (4.14 g, 30.0 mmol, 1.2 equiv.) was added to the colourless soln. of 14 (10.72 g, 25.0 mmol) in THF (625 ml) and H₂O (250 ml) [8]. After 1 h stirring at r.t., the soln. turned yellow. TLC Control (CH₂Cl₂): $R_{\rm f}$ 0.78 (14), 0.32 (15), 0.23 (16). After a total of 6 h stirring at r.t., H₂O (500 ml) was added, the mixture extracted with AcOEt (3 × 200 ml), and the org. phase washed with H₂O (200 ml), dried (MgSO₄), and evaporated. Careful chromatography of the dark-brown smelly oil (13.25 g; 15/16 8:6, $t_{\rm R}$ 32.3 (15), 30.6 min (16), SFC) on silica gel (1 kg, hexane/AcOEt 98:2 \rightarrow 97:3; TLC (hexane/AcOEt 3:1); $R_{\rm f}$ 0.47 (15), 0.43 (16)) gave first 5.83 g (44%) of 15 as a pale yellow viscous oil (96.1% chemical and 99.8% diastereoisomeric purity³),

²) Caution: heating (vacuum distillation, 250°) of this compound results in epimerization at C(2).

³⁾ Chemical purity means content of 15/16, diastereoisomeric purity corresponds to the ratio of the two stereoisomers 15 and 16 in this procedure.

SFC), then 1.17 g (9%) of oily 15/16 (ca. 1:3), and then 3.69 g (28%) of 16 as yellowish tacky crystals (97.3% chemical and 99.2% diastereoisomeric purity³). Total yield 81%. Pure 16 was obtained by recrystallization of crude 16 (9.5 g; from several batches) from pentane (150 ml): 100% chemical and 99.5% diastereoisomeric purity³).

Data of 15 (from another experiment which gave material of 95.4 chemical and 100% diastereoisomeric purity³)): $[\alpha]_{389}^{29} = +14.0$ (c = 1.0, CH₂Cl₂). IR (film): 3278s (OH), 2926s, 1463s, 1422s, 1379m, 1266m, 1238m (arylether), 1127s, 1102s (OH), 811w. ¹H-NMR (250 MHz, CDCl₃): 0.85 (d, J = 6.5, MeCH); 0.87 (d, J = 6.5, 2 MeCH); 0.88 (d, J = 6.5, MeCH); 0.96–1.62 (m, 19 aliph. H); 1.32 (s, Me-C(2)); 1.84 (t, J = 8, CH₂(1')); 2.09, 2.13 (2s, Me-C(7), Me-C(8)); 2.41 (d, J = 5, OH); 3.79 (s, MeO); 4.19 (d, J = 9, CHBr); 4.94 (dd, J = 9, 5, CHOH); 6.81 (s, arom. H). MS: 526, 524 (18, 17, M⁺), 444 (10, [M - HBr]⁺), 4.15 (13, [M - HBr - CHO]⁺), 283, 281 (3, 3, [M - C₁₆H₃₃ - H₂O]⁺), 219 (11, [M - C₁₆H₃₃ - HBr]⁺), 203 (33, [M - C₁₆H₃₃ - HOBr]⁺), 181 (100), 180 (C₁₀H₁₂O₃), 165 (21, C₁₀H₁₃O₂), 153 (13). Anal. calc. for C₂₉H₄₉BrO₃ (525.61): C 66.27, H 9.40, Br 15.20; found: C 66.14, H 9.55, Br 15.24.

Data of 16: Pale yellow crystals. M.p. 75.0° . [α] $_{889}^{25}$ = +6.6 (c = 1.0, CH₂Cl₂). IR (nujol): 3391s (OH), 2925s, 1466s, 1380m, 1268m, 1234m, 1127s, 1104m (OH); 838w. 1 H-NMR (250 MHz, CDCl₃): 0.80 (d, J = 6.5, MeCH); 0.82 (d, J = 6.5, MeCH); 0.86 (d, J = 6.5, 2 MeCH); 0.90–1.45 (m, 19 aliph. H); 1.45–1.65 (m, 1.45–1.65 (m, 11.45–1.65 (m, 11.55–1.65 (m, 11.55–1.55 (m, 11.55); 11.550, 11.551, 11

Recrystallization of a sample of 16 from MeOH/H₂O gave colourless long needles for X-ray analysis. M.p. 76.0°

(2R,3R)-3,4-Dihydro-6-methoxy-2,7,8-trimethyl-2- $\int (4R,8R)$ -4,8,12-trimethyltridecyl $\int 2H-1$ -benzopyran-3ol (18). 1m LiEt, BH in THF (36 ml, 3 equiv. of hydride) was added dropwise into the stirred soln. of 15 (6.31 g, 12.0 mmol; i.e. 6.61 g of 95.4% material) in THF (100 ml) at -78° during 20 min. The soln. was allowed to warm up to r.t. overnight (16-20 h), and H₂O (5 ml) and AcOH (10 ml) were added subsequently to hydrolyze the intermediates (boric esters/complexes, detected by NMR and MS in isolated products from previous experiments). After stirring for 1 h at r.t., the mixture was diluted with Et₂O (200 ml), the org. phase washed with 2N NaOH (2 × 100 ml) and H₂O (100 ml), the aq. phase extracted with Et₂O (150 and 100 ml), the combined org. extract dried (MgSO₄) and evaporated, and the crude oil (ca. 6.5 g) chromatographed (silica gel (800 g), hexane/AcOEt 19:1→9:1; TLC (hexane/AcOEt 3:1): $R_{\rm f}$ 0.47 (15), 0.35 (18)): 5.11–5.36 g (95–100 %; 1-mmol experiments gave 70–81 %) of 18 as a colourless oil; chemical purity 99.5-99.8% (GLC), diastereoisomeric purity at C(3) > 99.7% (HPLC, 3.5% AcOEt +0.2% i-PrOH in hexane). $[\alpha]_{589}^{20} = -0.71$ (c = 0.7, CH₂Cl₂). IR (film): 3361s (OH), 2926s, 1464s, 1423m, 1378m, 1237m (aryl ether), 1126m, 1098m, 1056w, 829w. UV (hexane): 291 (3700). H-NMR (250 MHz, CDCl₃): 0.84, 0.87 (2d, 4 MeCH); 0.96–1.62 (m, ca. 19 aliph. H); 1.19 (s, Me-C(2)); 1.69 ('t', J = 7.5, $CH_2(1')$); 1.74 (s, OH); 2.12, 2.13 (2s, Me-C(7), Me-C(8)); 2.74 (dd, H_R -C(4)); 3.05 (dd, H_R -C(4)); 3.75 (s, MeO); 3.80 ('t', CHOH); 6.40 (s, arom. H); $J(3,4\alpha) = 4.7$, $J(3,4\beta) = 4.2$, $J(4\alpha,4\beta) = 17.0$. MS: $446(24, M^+)$, $428(8, [M - H_2O]^+)$, 413 (2, $[M - H_2O - Me]^+$), 203 (65, $[M - C_{16}H_{33} - H_2O]^+$), 165 (100, $[C_{10}H_{13}O_2]^+$). Anal. calc. for $C_{29}H_{50}O_3$ (446.72): C 77.97, H 11.28; found: C 77.82, H 11.49.

(2 R,3 R)-3,4-Dihydro-6-methoxy-2,7,8-trimethyl-2-[(4 R,8 R)-4,8,12-trimethyltridecyl]-2H-1-benzopyran-3-yl Trifluoromethanesulfonate (19). Trifluormethanesulfonic anhydride (0.20 ml, 339 mg, 1.2 mmol) was added to the stirred soln. of 18 (447 mg, 1.0 mmol) and 2,6-dimethylpyridine (0.17 ml, 161 mg, 1.5 mmol) at −78° (→ white precipitate). The mixture was allowed to warm up to 0° and stirred for 1 h (TLC control (hexane/AcOEt 9:1): R_f 0.49 (19), 0.10 (18)), diluted with cold Et₂O (100 ml), and washed in turn with 2n H₂SO₄, sat. NaHCO₃ soln., and H₂O (50 ml of 0° each). The aq. phases were extracted with Et₂O (50 ml) and the combined org. extracts dried (MgSO₄) and evaporated (temp. < 20°). The crude turbid colourless oil (560–600 mg, ca. 100%) was characterized and used without further purification in the next step. Attempted column hromatography (silica gel), workup at higher temp., and standing at r.t. led to rapid decomposition. IR (film): 2927s, 1465m, 1414s, 1245m, 1211s (SO₂), 1146s, 914s, 609w. ¹H-NMR (250 MHz, CDCl₃): 0.84 (d, $J \approx 6.5$, MeCH); 0.85 (d, $J \approx 6.5$, MeCH); 0.87 (2d, $J \approx 6.5$, $J \approx 6.5$

 $(2\,\mathrm{R},3\,\mathrm{R})$ -3,4-Dihydro-6-methoxy-2,7,8-trimethyl-2-[(4 $\mathrm{R},8\,\mathrm{R})$ -4,8,12-trimethyltridecyl]-2H-1-benzopyran-3-yl Toluene-4-sulfonate (20). The soln. of 18 (447 mg, 1.0 mmol), $\mathrm{Et}_3\mathrm{N}$ (0.42 ml, 304 mg, 3.0 mmol), and TsCl (381 mg, 2.0 mmol) in pyridine (20 ml) was refluxed for 3 h (TLC control (hexane/AcOEt 9:1): R_f 0.66 (14), 0.23 (20), 0.08 (18)). Additional TsCl (191 mg, 1.0 mmol) was added and heating continued for 9 h. The dark-brown mixture

was diluted with CH₂Cl₂ (50 ml), washed successively with 2n H₂SO₄, sat. NaHCO₃ soln., and H₂O (100 ml each), the aq. phase extracted with CH₂Cl₂ (50 ml), and the combined org. extract dried (MgSO₄) and evaporated. Column chromatography (silica gel (200 g), hexane/AcOEt 9:1) afforded, after a fraction containing 14, 393 mg (65%) of 20 as a colourless oil which crystallized upon standing: off-white crystals. M.p. 84-85°. Purity 98.6% (SFC). Data given below originate from this material. [α] $_{589}^{20}$ = -16.2 (c = 1.0, CH₂Cl₂). IR (KBr): 2924s, 1596w, 1465m, 1366m, 1175s (aryl-SO₂), 1099m, 896s, 814m (p-disubst. benzene), 738m, 664m, 554m. ¹H-NMR (250 MHz, CDCl₃): 0.81 (d, MeCH); 0.84 (d, MeCH); 0.87 (2d, 2 MeCH); 0.92-1.64 (m, ca. 21 aliph. H); 1.13 (s, Me-C(2)); 2.07, 2.09 (2s, Me-C(7), Me-C(8)); 2.45 (s, MeC₆H₄); 2.94 (dd, H_B-C(4)); 3.06 (dd, H_A-C(4)); 3.72 (s, MeO); 4.63 (dd, CH-O); 6.28 (s, arom. H); 7.33 (XX' of AA'XX', 2 H); 7.79 (AA' of AA'XX', 2 H); J(3,4A) = 6, J(3,4B) = 7, J(4A,4B) = 17. MS: 600 (14, M^+), 428 (22, [M - TsOH] $^+$), 413 (2, [M - TsOH - Me] $^+$), 203 (100, [M - C₁₆H₃₃ - TsOH] $^+$), 165 (32, [C₁₀H₁₃O₂] $^+$), 91 (16, [C₇H₇] $^+$). Anal. calc. for C₃₆H₅₆O₅S (600.90): C 71.96, H 9.39, S 5.34; found: C 71.77, H 9.26, S 5.40.

Recrystallization of a sample of 20 from MeOH gave colourless, thin plates for X-ray analysis. M.p. 85.5-86.0°.

Single-Crystal X-Ray Analyses of 16 and 20^4). The intensity data for the single-crystal X-ray analyses were measured on an Enraf-Nonius-CAD-4 diffractometer (graphite-monochromated CuK_x radiation, ω -2 θ scans). The data were corrected for absorption. The structures were solved by a multiple-solution procedure [14] and refined by full-matrix least squares. In the final refinement, the non-H-atoms were refined anisotropically. The H-atoms were included in the structure-factor calculations, but their parameters were not refined. The absolute configurations of 16 and 20 were based on the anomalous scattering of the Br- and S-atom, resp., and were established by refining both enantiomers in each case. The final weighted R values were 0.0559 and 0.0708, resp., for the configurations shown in Figs. 2 and 3, and 0.0581 and 0.0731 for their corresponding antipodes. Thus, by Hamilton's test [15], the configurations shown in Figs. 2 and 3 are absolute.

Bromoalcohol 16 crystallized in the monoclinic space group $P2_1$ with a=5.529(1), b=49.866(4), c=11.039(1) Å, $\beta=103.98(1)^\circ$, V=2953.3(6.3) Å³, Z=4, $d_{\rm calc}=1.182$ g·cm⁻³, μ (Cu K_x) = 20.8 cm⁻¹. The cell unit contained two independent molecules, *i.e.* two molecules not related by crystallographic symmetry. The size of the crystal used for data collection was $ca.0.04 \times 0.08 \times 0.39$ mm. Of the 4444 independent reflections for $\theta<60^\circ$, 3181 were considered observed (I>2.0 σ (I)). Three reflections which were strongly affected by extinction were excluded from the final refinement and difference map. The final discrepancy indices were R=0.058 and wR=0.056 for the 3178 observed reflections. The major peaks (<0.6 e·Å⁻³) of the final difference map are near the Br-atoms.

Tosylate 20 crystallized in the monoclinic space group $P2_1$ with a=10.005(7), b=8.667(2), c=20.004(14) Å, $\beta=93.76(6)^\circ$, V=1730.8(6.7) Å³, Z=2, $d_{\rm calc}=1.153$ g·cm⁻³, μ (Cu K_a) = 10.9 cm⁻¹. In the crystal, most of the alkyl chain was disordered. Perspective drawings of rotamers A and B are shown in Fig. 3. The crystal used for data collection, $ca.0.06\times0.22\times0.42$ mm, was cooled with a N₂ gas stream to 115 K. Of the 3809 independent reflections for $\theta<75^\circ$, 3078 were considered observed ($I>3.0~\sigma(I)$). Twelve reflections which were strongly affected by extinction were excluded from the final refinement and difference map. The final discrepancy indices were R=0.060 and wR=0.071 for the 3066 observed reflections. The major peaks ($<1.1~\rm e^+Å^{-3}$) of the final difference map were near the S-atom.

Details of the crystal-structure investigations (table of final atomic parameters, anisotropic thermal parameters, bond distances, and angles) are available on request from the Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, England.

Treatment of tosylate 20 with 20 equiv. of LiEt₃BD (5 d, reflux) gave, after identical workup, 32% of 14, 8% of (3-²H)-12, 3% of starting material 20, and 51% of 18.

Data of $(3^{-2}H_1)$ -12 (3S): colourless oil. ¹H-NMR (400.1 MHz, CDCl₃): 0.84 (d, J = 6.6, MeCH); 0.85 (d, J = 6.4, MeCH); 0.86 (d, J = 6.3, 2 MeCH); 0.96–1.62 (m, ca. 21 aliph. H); 1.24 (s, Me–C(2)); 1.77 ('t', J(3,4 α) + J(3,4 β) = 14, H $_{\beta}$ –C(3)); 2.11, 2.12 (2s, Me–C(7), Me–C(8)); 2.69 (AB, J = 6.5, 16, H $_{\beta}$ –C(4)); 2.73 (AB, J = 7.5, 16, H $_{\alpha}$ –C(4)); 3.75 (s, MeO); 6.42 (s, arom. H). MS: 431 (48, M⁺), 430 (2, M⁺ undeuterated), 206 (10, M–C₁₆H₃₃]⁺), 165 (100, M–C₁₆H₁₃O₂]⁺), 164 (29, M–C₁₀H₁₂O₂]⁺); D-content > 95%.

(2R,4R)-3,4-Dihydro-6-methoxy-2,7,8-trimethyl-2-[(4R,8R)-4,8,12-trimethyltridecyl]-(4-2H)-2H-1-benzo-pyran ((4R)-(4-2H)-12). Obtained from 15 in 12% yield by consecutive treatment with 1) LiEt₃BD, 2) Tf₂O/2,6-dimethylpyridine, and 3) LiEt₃BH, as described above for (3S)-(3-2H)-12. Colourless oil. ¹H-NMR (400.1 MHz, CDCl₃): 0.84 (d, J = 6.5, MeCH); 0.85 (d, J = 6.3, MeCH); 0.87 (d, J = 6.5, J MeCH); 0.96-1.63 (M, J = 6.10 (M, J = 6.10 (M, J = 6.10 (M, J = 6.10 (M); 2.68 (J = 6.10 (M); 3.75 (J M); 0.42 (J = 6.15 (J M); 0.42 (J = 6.15 (J M); 0.43 (J = 6.15 (J M); 0.45 (J M); 0.46 (J M); 0.47 (J M); 0.47 (J M); 0.49 (J

(2R, 3R, 4S)-3,4-Dihydro-6-methoxy-2,7,8-trimethyl-2-[(4R,8R)-4,8,12-trimethyltridecyl]-(3,4- 2H_2)-2H-1-benzopyran ((3R,4S)-(3,4- 2H_2)-12). Obtained from 16 in 15% yield by consecutive treatment with 1) LiEt₃BD, 2) Tf₂O/2,6-dimethylpyridine and 3) LiEt₃BD, as described above for (3S)-(3- 2H)-12. Colourless oil. 1H -NMR (400.1 MHz, CDCl₃): 0.84 (d, J = 6.5, MeCH); 0.85 (d, J = 6.3, MeCH); 0.87 (d, J = 6.4, 2 MeCH); 0.95-1.62 (m, ca. 21 aliph. H); 1.24 (s, Me-C(2)); 1.70 (d, J(3a,4a) = 6.2, H_a -C(3)); 2.11, 2.12 (2s, Me-C(7), Me-C(8)); 2.70 (d, H_a -C(4)); 3.75 (s, MeO); 6.42 (s, arom. H). MS: 432 (67, M^+), 431 (5, M^+ monodeuterated), 207 (10, $[M-C_{16}H_{33}]^+$), 166 (100, $[C_{10}H_{12}DO_2]^+$), 165 (32, $[C_{10}H_{11}DO_2]^+$); D₂-content 93%, D₁-content 7%.

(2RS,4'R,8'R)-2-ambo-γ-Tocopherol ((2RS,4'R,8'R)-1). To a stirred soln. of 2,3-dimethylhydroquinone (13.8 g, 0.1 mol) in HCOOH (70 ml) was added over 2 h at 50° natural trans-phytol (29.65 g, 0.1 mol). The mixture was refluxed for 2.5 h, cooled to r.t., transferred to a round-bottom flask using toluene (100 ml), and concentrated at 45° under reduced pressure. The concentrate was diluted with MeOH (130 ml) and 5N H₂SO₄ (27 ml), and the mixture refluxed for 2 h and cooled to r.t. After the addition of hexane (200 ml), the mixture was extracted 3 times with ca. 20 ml each of 80% aq. MeOH. The MeOH phases were discarded and the upper layers combined, dried (MgSO₄), and evaporated. The dark brown oil (41.2 g) was chromatographed (silica gel (1 kg), hexane/Et₂O 9:1): 31.7 g (ca. 75%) of (2RS,4'R,8'R)-1. Red-brown oil. Purity 98.8% (GLC). This material was used directly for the next step (preparation of the camphanates 24/25).

5-[(7' R,11' R)-2',3'-Epoxyphytyl]-2,3-dimethyl-1,4-hydroquinone Diacetate (= 5-[(7' R,11' R)-2',3'-Epoxyphytyl]-2,3-dimethylbenzene-1,4-diyl Diacetate; **21/22**). To a 2-phase system consisting of CH₂Cl₂ and sat. Na₂CO₃ soln. (1 ml each) were added at 0° 80% 3-chloroperbenzoic acid (27 mg, 0.13 mmol) and **5** (45 mg, 0.09 mmol; obtained by acetylation of **2** [1]). The cooling bath was removed and the mixture stirred at r.t. After 20 min, the mixture was extracted with CH₂Cl₂(3 × 10 ml), the combined org. phase dried (MgSO₄) and evaporated, and the residue purified on 2 anal. TLC plates (Merck, F_{254} , 20 by 20 cm, hexane/AcOEt 8:2): **21/22** (38 mg, 82%). Colourless oil. ¹H-NMR (300 MHz, CDCl₃): 6.78 (s, H-C(6)); 2.83 (t, ³J = 5.88, H-C(2')); 2.69 (dd, ²J(1a',1b') = 15.10, ³J(1a',2') = 5.92, H_a-C(1')); 2.59 (dd, ²J(1a',1b') = 15.10, ³J(1b',2') = 5.59, H_b-C(1')); 2.29, 2.24 (2s, 2 Ac); 2.00 (s, Me-C(3), Me-C(4)); 1.44-0.98 (m, 25 H); 0.80-0.75 (m, Me-C(7'), Me-C(11'), Me-C(15'), Me(16')). CI-MS: 536 (7), 535 (37), 534 (100, [M + NH₄]⁺), 409 (6), 408 (23), 330 (7), 308 (7), 296 (8), 262 (16), 69 (9).

5-[(3' RS, 7'R, 11' R)-3'-Hydroxyphytyl]-2,3-dimethyl-1,4-benzoquinone ((3'RS)-23). To a soln. of 21/22 (5 mg, 9.7 µmol) in dry Et₂O (1 ml) was added LiAlH₄ (2 mg, 53 µmol) under Ar and stirred at 0° for 2 h. The reaction was then quenched with ice (1 g) and sat. Seignette-salt soln. (1 ml) and stirring continued for a further 10 min. Extraction of the mixture with Et₂O (2 × 10 ml), drying of the combined org. layers, evaporation, purification of the residue on 1 anal. TLC plate (Merck, F_{254} , 20 by 20 cm, hexane/AcOEt 8:2), and oxidation by air furnished 3.5 mg (83%) of (3'RS)-23. Yellow oil. 1 H-NMR (300 MHz, CDCl₃): 6.46 (s, H-C(6)); 2.42 (m, 2 H-C(1')); 1.96, 1.94 (2s, Me-C(3), Me-C(4)); 1.57-0.97 (m, 26 H); 0.81-0.76 (m, Me-C(7'), Me-C(11'), Me-C(15'), Me(16')). EI-MS 432 (0.8, M^{++}), 415 (1.3, $[M-OH]^{+}$), 207 (17), 164 (66), 43 (100).

The labelled compounds $(2'S,3'R)-(2'-^2H)-23/(2'R,3'S)-(2'-^2H)-23$ were prepared by using LiAlD₄ and transformed further by 1) Na₂S₂O₄ treatment (stirring in CHCl₃/EtOH/H₂O), 2) cyclization with a cat. amount of TsOH [2], 3) separation *via* the camphanates (*vide infra*), 4) LiAlH₄ treatment in Et₂O, and 5) methylation (Mel, NaH, THF) to finally yield $(3S)-(3-^2H)-12$ (3.0 mg, 65% based on 23).

Camphanate 24 of y-Tocopherol. A soln. of natural y-tocopherol (1; 58 mg, 0.14 mmol) and (-)-camphanoyl chloride (45 mg, 0.21 mmol, 1.5 equiv.) in dry pyridine (2 ml) was stirred at r.t. overnight. The mixture was then

diluted with Et₂O (50 ml) and washed with 2M HCl (3 × 50 ml). The org. layer was dried (MgSO₄), evaporated, and purified by FC (SiO₂, hexane/AcOEt 8:2): **24** (88 mg, 0.14 mmol, 100%). Colourless solid. Pure by TLC (hexane/Et₂O 4:1, R_f 0.14; CH₂Cl₂/hexane 9:1 (2 times), R_f 0.30) and HPLC (*Fig.* 4). M.p. 52–55°.

Camphanates 24/25 were prepared by the same procedure from 2-ambo-y-tocopherol (2-ambo-1). TLC (hexane/Et₂O 4:1): 1 spot. TLC (CH₂Cl₂/hexane 9:1 (2 times)): $R_{\rm f}$ 0.30 (24), 0.27 (25); most conveniently, 24/25 was separated by HPLC (Fig. 4).

Data of 24: 1 H-NMR (300 MHz, CDCl₃): 6.51 (s, H-C(5)); 2.64 (t, ${}^{3}J$ = 6.71, 2 H-C(4)); 2.54-2.45 (m, camph.); 2.18-2.09 (m, camph.); 2.04, 1.97 (2s, Me-C(7), Me-C(8)); 1.95-1.87 (m, camph.); 1.78-1.60 (m, camph., 2 H-C(3)); 1.46-0.97 (m, 20 H); 0.81-0.76 (m, Me-C(4'), Me-C(8'), Me-C(12'), Me(13')). CI-MS: 615 (33), 614 (100, [M + NH₄] $^{+}$), 597 (5, [M + H] $^{+}$), 596 (8), 434 (9), 417 (8).

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